BCH202: INTRODUCTION TO PHYSICAL BIOCHEMISTRY COURSE LURTURERS: Dr. (Mrs.) M.O. LAMBE and Mr. A. BALOGUN



جامعة الحكمة، إلورن-نيجيريا

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FACULTY OF NATURAL AND APPLIED SCIENCES

DEPARTMENT OF BIOLOGICAL SCIENCES

B.Sc. BIOCHEMISTRY PROGRAMME 2019/2020 ACADEMIC SESSION

B. Sc. BIOCHEMISTRY PROGRAMME

COURSE OUTLINE FOR

COURSE CODE: BCH 202

COURSE TITLE: INTRODUCTION TO PHYSICAL BIOCHEMISTRY

CREDIT UNIT: 3

26 LECTURE PERIODS: Mondays, 11 am – 1 pm; Wednesdays 3 – 4 pm

STUDENTS SHOULD NOTE THAT AFTER EVERY TOPIC THERE SHALL BE A SHORT

CONTINUOUS ASSESSMENT TEST ON GOOGLE CLASSROOM PLATFORM

LECTURE	DATE	TOPIC	COURSE
PERIOD			LECTURER
1	24th	Water	Dr. Lambe, M.O.
	February,	Physical Properties	
	2020	Structure of water molecules	
		As Universal Solvent	
2	26th	Water	Dr. Lambe, M.O.
	February,	Weak Interactions in Aqueous Solutions	
	2020	 Van der Waals Interactions 	
3	2nd March,	Solutions	Dr. Lambe, M.O.
	2020	 Definition 	
		 Types/forms and their preparations 	
4	4th March,	Solutions	Dr. Lambe, M.O.
	2020	 Forms of Expressing Solution 	
		Concentration	
		 Inter- conversion of the various form of 	
		Solution Concentration Expression	
		 Solution Preparation 	
5	9th March,	pH and Buffer	Dr. Lambe, M.O.
	2020	 Ionization of water, weak acids and weak 	
		bases	
		pH Scale	
		Bronsted-Lowry Concept of Acid and Base	
6	11th	pH and Buffer	Dr. Lambe, M.O.
	March,	 Titration of a weak acid by a strong base 	
	2020	 Henderson-Hasselbach Equation 	
7	16th	Buffer and Biological Buffer Systems	Dr. Lambe, M.O.

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	March,	Henderson-Hasselbach Equation	
	2020	The Phosphate Buffer System	
		The Bicarbonate Buffer System	
8	18th	Biological Buffer Systems	Dr. Lambe, M.O.
	March,	The Protein Buffer System	
	2020	The Amino acids Buffer System	
		The Heamogblobin Buffer System	
9	23rd	Donnan equilibrium	Mr. Balogun, A.
	March,	- Definition and Characteristic features	Time Date gam, Th
	2020	- Assumptions, Calculations and Applications	
10	25th	Donnan equilibrium	Mr. Balogun, A.
	March,	- Definition and Characteristic features	, , , ,
	2020	- Assumptions, Calculations and Applications	
11	30th	Chemical Kinetics	Mr. Balogun, A.
	March,	- Definition and Characteristic features	,
	2020	- Rate law of Reaction orders	
12	1st April,	Chemical Kinetics	Mr. Balogun, A.
	2020	- Integrated Rate Laws: Concentration	
		Changes over Time	
		- Reaction Half-Life of Reaction orders	
13	6th April,	Chemical Kinetics	Mr. Balogun, A.
	2020	- Reaction Half-Life of Reaction orders	
		- The Effect of Temperature on the Rate	
		Constant and the Rate	
14	8th April,	Chemical Equilibrium	Mr. Balogun, A.
	2020	 Definition and Basic concept 	
		- Features of Biochemical reactions	
15	13th April,	Applied Thermodynamics	Mr. Balogun, A.
	2020	- Basic concepts of thermodynamics	
		- Laws of thermodynamics	
16	15th April,	Applied Thermodynamics	Mr. Balogun, A.
	2020	- Free energy and equilibrium	
		- temperature dependence of equilibrium	
		constant	
17	20th April,	Redox reactions	Mr. Balogun, A.
	2020	- Basic concepts	
10	22 1 4 11	- free energy changes in redox reactions	76.51
18	22nd April,	Redox reactions	Mr. Balogun, A.
	2020	- Basic concepts	
10	2745 Amil	- free energy changes in redox reactions Electrochemical cells	Mr. Dologue A
19	27th April,	- Half-cell electrode potentials	Mr. Balogun, A.
	2020	<u> </u>	
		- Derivation and Applications of Nernst	
20	2041- 41	equation Floatrophomical calls	Ma Dologue A
20	29th April,	Electrochemical cells	Mr. Balogun, A.
	2020	- Half-cell electrode potentials	
		- Derivation and Applications of Nernst	
		equation	

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21	4th May, 2020	Final C.A. Test	
22	6th May, 2020	Final C.A. Test	
23	11th May, 2020	Final C.A. Test	
24	13th May, 2020	Final C.A. Test	
25	18th May, 2020	Revision Week	
26	20th May, 2020	Revision Week	